

## Subject card

Subject name and code	English in chemistry, PG_00081980								
Field of study	Chemistry								
Date of commencement of studies	October 2025		Academic year of realisation of subject			2027/2028			
Education level	Bachelor's studies		Subject group			Obligatory subject group in the field of study			
						Subject group related to scientific research in the field of study			
Mode of study	full-time studies		Mode of delivery			at the university			
Year of study	3		Language of instruction			Polish			
Semester of study	5		ECTS credits			1.0			
Learning profile	academic		Assessment form			credit			
Conducting unit	Faculty of Chemistry -> Rector								
Name and surname	Subject supervisor	prof. dr hab. inż. Marek Kwiatkowski							
of lecturer (lecturers)	Teachers								
Lesson types	Lesson type	Lecture	Tutorial	Laboratory	ratory Project		Seminar	SUM	
	Number of study hours	0.0	15.0	0.0	0.0		0.0	15	
	E-learning hours inclu	ıded: 0.0							
Learning activity and number of study hours	Learning activity	Participation in didactic classes included in study plan		Participation in consultation hours		Self-study		SUM	
	Number of study hours	15		2.0		8.0		25	
Subject objectives	To familiarize students with the general professional terminology used in English chemical texts. To introduce students to understanding professional publications on chemistry in English and to formulating simple chemical texts in this language.								
Learning outcomes	Course out	Subject outcome			Method of verification				
	[CHEML3_U12] Reads with understanding scientific and popular science chemical texts in English.		Student reads, understands and analyzes written chemical texts in English.			[SU1] oral statement/conversation/ discussion [SU4] test/exam - oral or written [SU8] observation of student's independent or team work			
	[CHEML3_U11] Prepares and presents oral presentations in various fields of chemistry in Polish and English, using acquired knowledge and skills as well as basic sources of scientific information.		Students presents orally selected chemical issues in English, discusses them using appropriate terminology.			[SU1] oral statement/conversation/ discussion [SU4] test/exam - oral or written [SU8] observation of student's independent or team work			
	[CHEML3_W05] Has basic knowledge of the chemical specialisation studied.		Student names English terms describing the basic concepts of chemistry.			[SW4] test/exam - oral or written [SW1] oral statement/ conversation/discussion			
	[CHEML3_U09] Is able to learn independently.		Student explores new chemical texts and solves tests veifying proper understanding of the content.			[SU1] oral statement/conversation/ discussion [SU4] test/exam - oral or written [SU8] observation of student's independent or team work			
	[CHEML3_U10] Prepares papers on various fields of chemistry in Polish and English, using acquired knowledge and skills as well as various sources of scientific information.		Student writes short chemical texts in English using appropriate terminology.			[SU1] oral statement/conversation/ discussion [SU4] test/exam - oral or written [SU8] observation of student's independent or team work			

Subject contents	Classification and naming of inorganic compounds. Classification and naming of organic compounds. Laboratory techniques and procedures, labware. Techniques, procedures and labware in chemical analysis.						
Prerequisites and co-requisites	Completing the courses "Inorganic chemistry" and "Organic chemistry"						
Assessment methods	Subject passing criteria	Passing threshold	Percentage of the final grade				
and criteria	Three short tests during classes.	51.0%	60.0%				
	Final test based on a selected chemical text	51.0%	40.0%				
Recommended reading	Basic literature	1. M. Kwiatkowski, P. Stepnowski "Język angielski w chemii i w ochronie środowiska", ed. Uniwersytet Gdański, Gdańsk 2010, electronic publication http://inf.ug.edu.pl/kierunkizamawiane/materialy/chemia/Angielski.pdf					
	Supplementary literature  1. Selected scientific publications on chemistry in English and fragments of original texts from English chemistry textbooks.						
	eResources addresses						
Example issues/ example questions/ tasks being completed	Give the name of the compound KH <sub>2</sub> PO <sub>4</sub> · 6H <sub>2</sub> O.						
	Give the systematic name and chemical classification of the compound CH <sub>3</sub> C(O)CH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OH.						
	Name the elements of the laboratory setup for fractional distillation under normal pressure.						
	Explain the term "magic nucleus".						
Work placement	Not applicable						

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